**Type II Ionic and Polyatomic Compounds #2**

*Please write the formula or name corresponding to the given name or formula for the following ionic compounds. You can use the “subscript”* *button in the “Home” bar to make small numbers.*

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| Chromium (II) chloride | PbF2 |
| Copper (I) sulphide | FeI2 |
| Iron (II) phosphide | CrBr2 |
| Manganese (II) oxide | Fe2O3 |
| Mercury (II) oxide | MnO |
| Tin (II) sulphide | Hg3N2 |
| Tin (IV) nitride | MnS |

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| --- | --- |
| Lead (IV) chloride | Sn3P4 |
| Chromium (III) chloride | FeI3 |
| Copper (I) iodide | CrBr3 |
| Sodium chromate | Al(OH)3 |
| Lithium dichromate | NaCH3COO |
| Magnesium hydroxide | Ba3(PO3)4 |
| Tin (II) hydroxide | Cr(OH)3 |
| Lead (II) hydroxide | (NH4)3PO4 |
| Manganese (IV) sulphate | Cr(CH3COO)3 |
| Potassium permanganate | Mg3(PO4)2 |